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From synchrotrons for XFELs: the soft x-ray near-edge spectrum of the ESCA molecule

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Abstract. A predictive understanding of soft x-ray near-edge absorption spectra of small molecules is an enduring theoretical challenge and of current interest for x-ray probes of molecular dynamics. We report the experimental absorption spectrum for the ESCA molecule (ethyl trifluoroacetate) near the carbon 1s absorption edge between 285-300 eV. The ESCA molecule with four chemically distinct carbon sites has previously served as a theoretical benchmark for photoelectron spectra and now for photoabsorption spectra. We report a simple edge-specific approach for systematically expanding standard basis sets to properly describe diffuse Rydberg orbitals and the importance of triple excitations in equation-of-motion coupled-cluster calculations of the energy interval between valence and Rydberg excitations.

1. Introduction

X-ray free-electron lasers (XFELs) [1, 2, 3, 4, 5, 6] and table-top sources of x-rays based upon high harmonic generation (HHG)[7, 8] have over the past decade energized the field of ultrafast atomic and molecular dynamics [9]. For XFELs, the intense tunable x-ray pulses that span wavelengths from ~ 40 to 0.5 \AA with sub-femtosecond pulse durations [10, 11, 12] and multi-pulse, multi-color options [13, 14, 15] allow one to investigate electronic and nuclear dynamics on their natural timescales. Using pump-probe schemes, one can not only study valence-excited dynamics, e.g. the mechanism of photoprotection of DNA bases via intersystem crossing [16], but also

inner-shell dynamics where competition between localized Auger decay and charge-transfer mediates dissociation in small molecules [17].

With the enormous versatility of pump and probe x-ray pulses, it is natural to contemplate the optimal observable for nuclear and electronic orbital dynamics [18] and even consider nonlinear multi-dimensional spectroscopy techniques [19, 20]. Two straightforward single-photon methods, photoelectron spectroscopy and photoabsorption spectroscopy are universally applicable. X-ray photoelectron spectroscopy will yield binding energies of core and valence orbitals and has been widely used for chemical analysis, as exemplified by the iconic ESCA molecule studies done by Siegbahn [21, 22]. The site-specificity of the four non-identical carbons is encoded in their binding energies which differ by ~ 8 eV from the electronegative CF_3 end to the CH_3 end, as shown in Fig. 1. A more recent combined experimental and theoretical study shows that the high-resolution C 1s photoelectron spectrum also possesses some limited sensitivity to the conformational state and nuclear dynamics associated with photoionization [23]. However, it was the intrinsic site-specific nature of the binding energies that motivated our search for chemical site-specificity in bond-breakage in the ESCA molecule using the photoelectron-photoion-photoion coincidence method [24]. We found limited site specificity. For all four carbon ionization sites, Auger decay weakens the same bonds and transfers the two charges to opposite ends of the molecule, which leads to a rapid dissociation into three fragments, followed by further fragmentation steps.

So the question naturally arises, what is the mechanism by which an initially localized 1s hole delocalizes across the molecule? Can we probe the evolution of the valence hole density across the molecule via snapshots of its near-edge x-ray absorption fine structure (NEXAFS) spectrum? Of course, NEXAFS has long been a powerful technique for chemical analysis [25] and has been more recently applied for time-resolved studies of evolving valence electronic structure using both HHG [26, 27] and XFEL sources [16]. Somewhat surprisingly, to date there is no published soft-x-ray absorption spectrum for the iconic ESCA molecule, ethyl trifluoroacetate. As our results show, the photoabsorption spectrum near the carbon K-edge is much more complicated than its photoelectron spectrum - with no one-to-one correspondence between the carbon site and the absorption spectrum (see Fig. 1).

Despite the complexity associated with near-edge absorption spectroscopy, there has been considerable progress in theoretical descriptions driven largely by advances in light source technology [28, 29]. Of particular relevance to the present work is the application of equation-of-motion coupled-cluster (EOM-CC) techniques [30, 31], which can be systematically converged towards the right answer, to calculations of core-level spectroscopy. The earlier work [32, 33, 34, 35, 36] shows promise of the EOM-CC methods for obtaining accurate core excitation energies, but was hampered by the difficulty of converging EOM-CC equations for core-excited states. With an elegant approach of introducing the core-valence separation (CVS) scheme [37] into the EOM-CC formulation [38], the CVS-EOM-CC methods have recently evolved into powerful tools for calculations of core ionization and excitation energies, photoelectron spectra, NEXAFS spectra, and resonant inelastic x-ray scattering (RIXS) spectra [39, 40, 41, 42, 43, 44, 45, 46, 47, 48, 49]. A unique appealing feature of CVS-EOM-CC is the capability of providing systematically improved results using the hierarchy of EOM-CC methods. Namely, while the popular EOM-CC singles and doubles (EOM-CCSD) method is capable of providing useful results [50, 43, 51, 52, 53], the inclusion of higher excitations in the CC hierarchy further enables systematic improvement of

computational results, which has the desired potential of paving the way to calculations that are essentially quantitative [42, 49, 45, 54, 55]. This has been exploited in the present study to assess the reliability of computational simulation for the fairly complicated NEXAFS spectrum of the ESCA molecule.

In this paper we report the experimental near-edge absorption spectrum of ethyl trifluoroacetate (the "ESCA" molecule) and theoretically describe the origin of its features using CVS-EOM-CC methods. Our theoretical developments include 1) a simple edge-specific approach for systematically converging basis set effects to properly describe Rydberg orbitals, and, 2) the demonstration of importance of including triple excitations to obtain accurate valence-to-Rydberg energy spacings in carbonyl containing compounds. These developments, which appear to be generally applicable, are benchmarked with calculations of simple small molecules, NH_3 , CO_2 , and CH_2O , and then used for the calculations of the NEXAFS spectrum for the ESCA molecule to facilitate the assignment of the experimental spectrum.

2. Experimental considerations

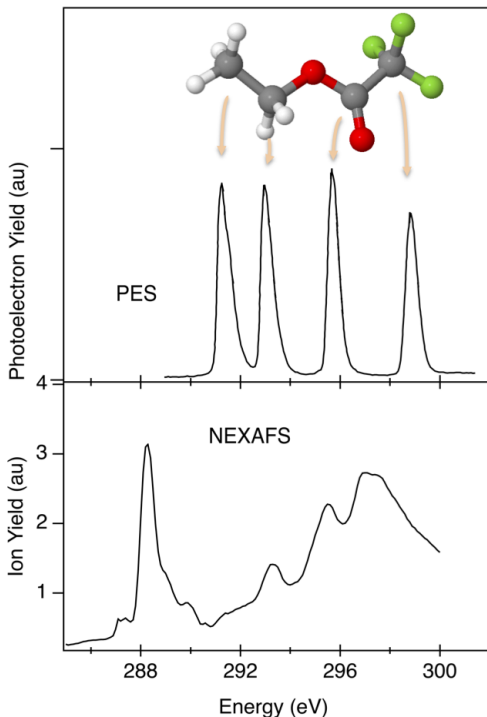


Figure 1: Comparison of photoelectron and photoabsorption spectra for the ESCA molecule. Top panel: the C 1s binding energy of ethyl trifluoroacetate recorded at a photon energy of 340 eV, extracted from [23]. There is a one-to-one correspondence between the peaks and the four distinct carbon atoms, with the 1s binding energy of

C_{CH_3} at 291.47 eV. Bottom panel: photoabsorption via total ion yield of ethyl trifluoroacetate obtained by scanning the photon energy from 285 to 300 eV (This work. See text for details).

Synchrotron measurements of absorption spectra of atoms and molecules have long contributed to the advances in theoretical methods to describe photoabsorption to bound and continuum states [56, 57, 58]. Relative to FELs the photon-molecule interactions are simpler, being constrained to single-photon interactions. The probability of photon absorption by an isolated single molecule is given by the product of the pulse fluence and its absorption cross section. For monochromatic synchrotron radiation, with a typical 10^6 photons per pulse focused to $1 \mu\text{m}^2$ and a cross section of 1 Mb, the probability of one photon being absorbed within a photon pulse is 10^{-4} and the probability for two-photon absorption 10^{-8} . The situation is vastly different at XFELs where one routinely obtains 10^{12} photons per pulse to easily saturate single-photon absorption and create multiphoton absorption [59, 60]. Detection of the single-photon x-ray absorption spectra in the gas phase at synchrotrons can be done either directly in transmission or by detection of secondary emitted particles, i.e. photons, electrons or ions. For the non-transmission measurements of photoabsorption, the observed spectral features are affected by molecular dynamics that control the branching ratios to the secondary products [25, 61]. The spectra reported here have been obtained via total-ion-yield spectroscopy which is closely related to photoabsorption [61].

The bottom panel of Fig. 1 shows the total-ion-yield spectrum of ethyl trifluoroacetate measured on the PLEIADES beam line at the SOLEIL synchrotron radiation facility. Ions were collected in a time-of-flight spectrometer with an extraction field of about 130 V/cm (200 V between the repeller and extractor plates, which are separated by 15.5 mm) [62, 63]. The photon-energy resolution in the measurement was 25 meV, considerably less than the lifetime width of a carbon 1s hole ~ 100 meV [64]. The photon energy was simultaneously calibrated using carbon dioxide in a separate ionization chamber and calibrated to the study by Adachi [65] with a precision of ~ 50 meV. The measurement was carried out using an effusive gas jet resulting in a background pressure of 5×10^{-7} mbar.

The C 1s total-ion-yield spectrum exhibits features corresponding to core electron excitation from each of the four chemically-shifted carbon sites. No previous absorption spectra have been reported in the region near the C 1s ionization thresholds. Several narrow absorption features are seen below the lowest carbon 1s ionization threshold, but the identification of these features and the character of the occupied valence orbitals, requires the detailed analysis described in Section 3.

3. Computational methods

The computational simulation of the NEXAFS spectra presented here has used core-valence separated equation-of-motion coupled-cluster (CVS-EOM-CC) [38] calculations for core excited states. In subsection 3.1, using NH_3 as a model system, we demonstrate the evolution of NEXAFS spectrum as a function of the basis set using an edge-specific approach to saturate basis set effects. Contributions from triple excitations are important for accurate calculations of NEXAFS for molecules containing a carbonyl group. Their effects are systematically studied for CH_2O at both C 1s and O 1s edges, and taken into account in the calculations of ESCA molecule, as detailed in subsection 3.2. We have correlated the targeted core orbital and have kept the other core orbitals frozen in ground-state CC calculations. This is also intrinsically an edge-specific scheme, consistent with the present edge-specific approach for treating basis-set effects on Rydberg states. It has been reported that the frozen core

version of CVS-EOM-CCSD (fc-CVS-EOM-CCSD) method [41] often yield accurate core excitation energies, benefiting from cancellation between errors of the frozen-core approximation and the neglect of triple excitations. However, since it is essential to include triple excitations to obtain accurate results for the ESCA molecule, it is necessary to correlate the 1s electrons of the targeted carbon in ground-state CC calculations. In all calculations presented here, scalar-relativistic effects have been taken into account using the spin-free exact two-component theory in its one-electron variant (SFX2C-1e) [66, 67] and the correlation-consistent basis sets [68] with SFX2C-1e recontraction. All calculations have been carried out using the CFOUR program package [69, 70, 30, 71, 72, 73]. The calculations presented here for the ESCA molecule have used the structure of the C_s conformer computed in Ref. [74]. We mention that similar results have been obtained for the C_1 conformer with details documented in Table 4 of the supplementary material.

EOM-CCSD density difference natural orbitals (DDNOs) have been used to enable intuitive understanding of the excitations in these core excited states. DDNOs are obtained by diagonalizing the one-electron density difference matrix, i.e., the difference between the excited state and ground state EOM-CCSD one-electron density matrices. For each core-excited state obtained in the present work, one obtains a set of DDNOs with one of them having an occupation number close to -1 (the 1s orbital excited from) and one having an occupation number close to 1 (the natural virtual orbital excited to). The latter has been plotted for twelve core excited states with the highest intensities in the NEXAFS spectrum of the ESCA molecule. For a first study using EOM-CCSD DDNOs to facilitate the analysis of NEXAFS spectra, we refer the reader to Ref. [40]. We also refer the reader to Ref. [75] as a recent perspective review on related concepts about transition natural orbitals for further discussions of orbital analysis in x-ray spectroscopy as well as Ref. [76] for DDNOs in other context.

3.1. An edge-specific approach for treating basis set effects on core excited Rydberg states

Core-excited Rydberg states are inadequately accounted for in calculations without using sufficient diffuse basis functions to capture the diffuse character of these states [43]. As shown in Fig. 2 for ammonia, the use of the standard cc-pVTZ basis [68] overestimates excitation energies for the $1s \rightarrow 3p$ states and misses all other Rydberg states. To solve this problem, we augment the standard correlation-consistent basis sets for the targeted atom systematically with diffuse s-, p-, and d-type functions. The exponents of diffuse functions are obtained by multiplying those of the most diffuse functions in the set by a geometric factor of 1/3. Interestingly, a convergence pattern has been observed. The addition of two sets of diffuse functions is required to saturate the description for the first set of Rydberg states, while each additional set of diffuse functions then serve to converge another set of Rydberg states. Namely, as demonstrated in Fig. 2, cc-pVTZ+2spd, cc-pVTZ+3spd, and cc-pVTZ+4spd sets provide converged results for $1s \rightarrow 3p$, $1s \rightarrow 4p$, and $1s \rightarrow 5p$ transitions, respectively. Therefore, in calculations of the ESCA molecule, the cc-pVTZ+4spd set has been adopted for the targeted carbon atom together with cc-pVTZ basis sets for all other atoms to ensure an accurate description of valence transitions and the first three sets of core excited Rydberg states.

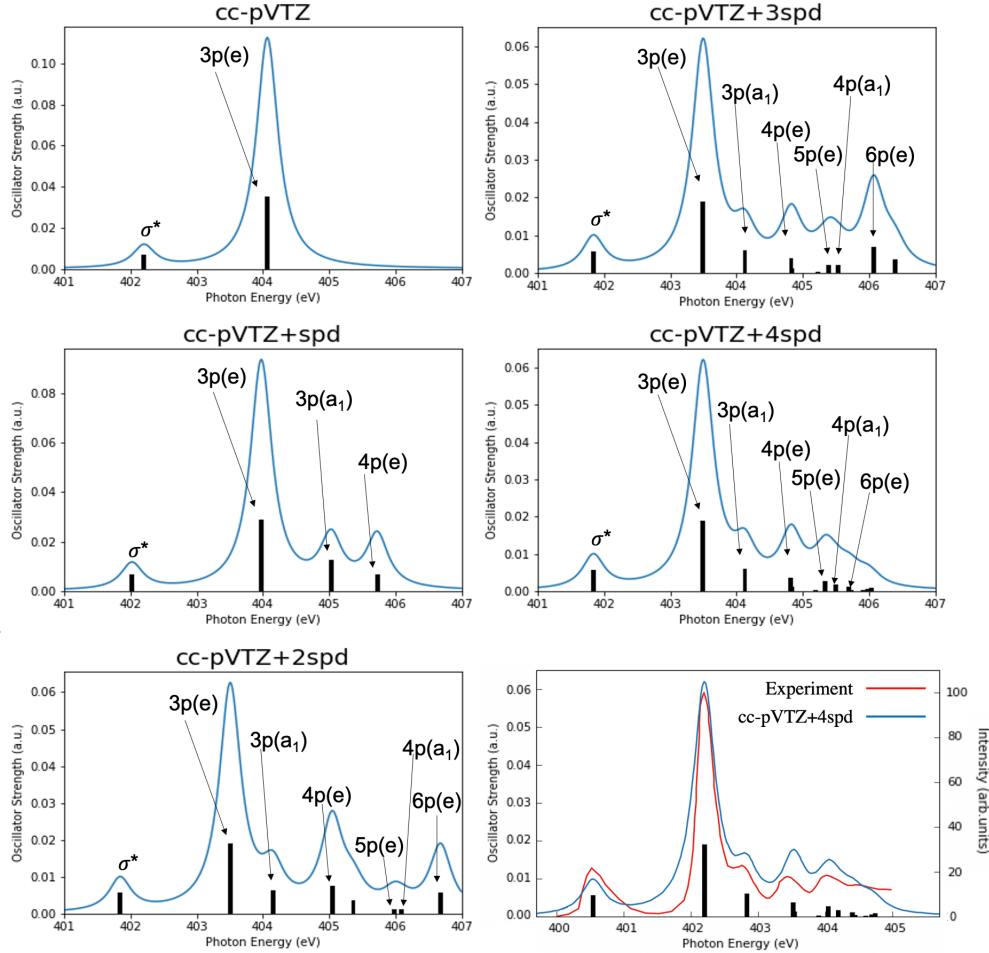


Figure 2: EOM-CCSD NEXAFS spectra for the nitrogen edge in ammonia obtained by convolution of the computed energies and oscillator strengths with a Lorentzian function with a FWHM value of 0.4 eV. Scalar-relativistic effects were taken into account using the spin-free exact two-component theory in its one-electron variant. "cc-pVTZ+xspd" (x=1, 2, 3, 4) refers to augmentation of nitrogen cc-pVTZ basis sets with x sets of additional diffuse s-, p-, and d-type functions. In the last figure, the black solid line represents experimental spectrum [77] and the computed spectrum has been red-shifted by 1.30 eV to align the computed $1s \rightarrow \sigma^*$ transition with the experiment.

The present scheme is edge specific in that a separate EOM-CC calculation is carried out for core excited states of each atom with additional diffuse functions located on this atom, e.g., in the case of the carbon NEXAFS of the ESCA molecule, each of the four carbon edges requires one separate EOM-CC calculation. It appears to be a useful alternative to the use of molecular Rydberg functions in Refs. [43]. The latter can be combined with the frozen core version of CVS-EOM-CC methods [41]

to enable calculations of core excited states of all edges with only a single ground-state CC calculation, being an edge-universal approach. On the other hand, for larger molecules, the present edge-specific approach for adding diffuse functions appears to be physically better motivated than using the center of charge as the center for molecular Rydberg functions, as the localized core hole serves as a well-defined center for core-excited Rydberg states.

3.2. Importance and treatment of triples contributions

The CVS-EOM-CC core ionization and excitation energies can be improved in a systematic way by increasing the rank of excitation operators [42]. The CVS-EOM coupled-cluster singles and doubles (CCSD) method has been shown to provide useful results for NEXAFS spectra [43, 51, 52]. Although EOM-CCSD calculations typically overestimate core ionization and excitation energies by 0.5-3 eV, the errors are often partially offset by the neglect of scalar-relativistic effects and correlation contributions of core electrons. Further, these errors tend not to contribute to relative shifts and, thus, do not affect the overall shape of the simulated spectra. For example, as already shown in Fig. 2, EOM-CCSD calculations describe the NEXAFS of ammonia quite well. However, a notable exception is that CVS-EOM-CCSD calculations have been reported to significantly overestimate the separation between the carbon (or oxygen) $1s \rightarrow \pi^*$ transition and the corresponding Rydberg transitions in the formaldehyde (CH_2O) molecule [43]. Here we show that the inclusion of triple excitations enables accurate calculations of these relative shifts. The carbon and oxygen $1s$ core excitation energies of CH_2O obtained from CVS-EOM-CCSD [30] and CC single doubles and triples (CCSDT) [78, 79] calculations are summarized in Table 1. Triples corrections (the difference between CCSDT and CCSD results) to the relative shifts between $1s \rightarrow \pi^*$ and the first Rydberg transitions amount to 0.5 eV for the carbon edge and as large as 0.8 eV for the oxygen edge. The inclusion of triples corrections reduces the separation between the $1s \rightarrow \pi^*$ transition and the first Rydberg transition from 4.68 eV to 4.24 eV for the carbon edge and from 5.28 eV to 4.44 eV for the oxygen edge. These are to be compared with the experiment values of 4.15 eV and 4.66 eV. As the ESCA molecule also contains a carbonyl group, it is therefore necessary to include triples corrections to obtain accurate results for its NEXAFS spectrum. The computational cost of CCSDT scales as the eighth power of the system size, and CVS-EOM-CCSDT calculations of the ESCA molecule are beyond our current computational resources. Therefore, we have adopted the EOM-CCSD(T)(a)* method [80] with a noniterative triples correction to EOM-CCSD, which has been shown to provide reasonable estimates for triples corrections to core excitation energies [49]. As shown in Table 1, the EOM-CCSD(T)(a)* values for relative shifts between the $1s \rightarrow \pi^*$ transition and the first Rydberg transition compare favorably with the EOM-CCSDT values, although EOM-CCSD(T)(a)* underestimates the triples corrections to the absolute values of these core excitation energies by around 0.4 eV.

Table 1: Carbon and oxygen K-edge excitation energies (eV) of formaldehyde. The relative shifts between the first Rydberg excitation $1s \rightarrow 3p(a_1)$ and the valence $1s \rightarrow \pi^*$ transition (eV) are enclosed in the parentheses. The cc-pVTZ+5spd basis set was used for the targeted atom and cc-pVTZ basis sets for the other atoms. Scalar-relativistic effects were taken into account using the spin-free exact two-component theory in its one-electron variant.

	carbon edge		oxygen edge	
	$1s \rightarrow \pi^*$	$1s \rightarrow 3p(a_1)$	$1s \rightarrow \pi^*$	$1s \rightarrow 3p(a_1)$
EOM-CCSD	286.59	291.27 (4.68)	532.49	537.77 (5.28)
EOM-CCSD(T)(a)*	286.28	290.59 (4.31)	531.56	536.10 (4.54)
EOM-CCSDT	285.90	290.14 (4.24)	531.03	535.47 (4.44)
Experiment [81]	285.97	290.12 (4.15)	530.88	535.54 (4.66)

4. Results and discussion

4.1. Assignment of experimental NEXAFS spectrum using computational results

Fig. 3 shows the computed NEXAFS spectrum for the four carbon edges of the ESCA molecule. The calculations have used the SFX2C-1e EOM-CCSD(T)(a)* method for transition energies, EOM-CCSD method for transition dipole moments, cc-pVTZ+4spd basis set for the targeted carbon, and cc-pVTZ basis sets for all other atoms. The density difference natural virtual orbital as plotted in Fig. 4-(A) shows that the first intense peak located at 288.80 eV in the calculation corresponds to the carbonyl carbon $1s \rightarrow \pi^*$ transition. Here this excitation into the valence π^* orbital leads to a transition energy around 8 eV lower than the $1s$ ionization energy of the carbonyl carbon, which lies in the middle of pre-edge transitions of the methyl carbon edge. Computed transition energies for $1s \rightarrow \pi^*$ excitations in CO_2 , CH_2O , and the ESCA molecule are summarized in Table 2. The transition energies obtained from EOM-CCSD(T)(a)* / cc-pVTZ+4spd calculations are consistently around 0.5 eV higher than the corresponding experimental values, perhaps mainly due to the approximation in the treatment of triple excitations. The intense feature peaked at 288.31 eV in the experimental spectrum can thus be confidently assigned to this carbonyl carbon $1s \rightarrow \pi^*$ transition. Further, in Fig. 3 we red-shifted all computed lines by 0.50 eV to align the computed and experimental peaks for this transition. In the following discussions we will also use the red-shifted computed energies.

Table 2: Carbon $1s \rightarrow \pi^*$ transition energies (eV). Scalar-relativistic effects were taken into account using the spin-free exact two-component theory in its one-electron variant.

	CO_2	CH_2O	ESCA
EOM-CCSD/cc-pVTZ+4spd	291.60	291.27	289.11
EOM-CCSD(T)(a)* / cc-pVTZ+4spd	291.33	290.59	288.80
EOM-CCSDT/cc-pVTZ+4spd	290.95	290.14	/
Experiment [81, 82]	290.76	290.12	288.31

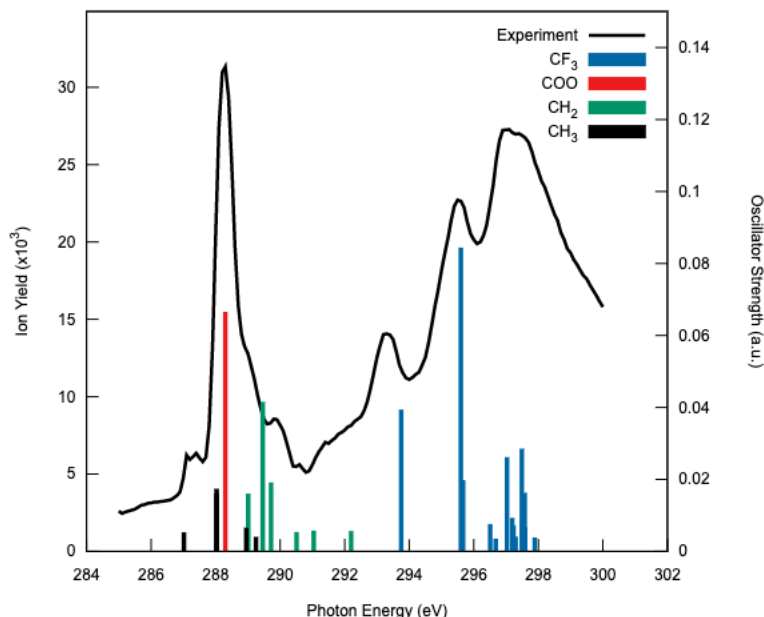


Figure 3: Experimental and computed NEXAFS spectra for the four carbon edges of the ethyl trifluoroacetate molecule. The black solid line shows the variation of measured ion yield with respect to photon energies. Blue, red, green, and black sticks represent computed absorption lines for the CF_3 , COO , CH_2 , and CH_3 carbon edges, respectively, with the height being the values of the computed oscillator strength. Energy level positions were computed using the EOM-CCSD(T)(a)* method, while transition dipole moments were obtained from EOM-CCSD calculations. The computed spectrum has been red-shifted by 0.50 eV to align the computed $1s \rightarrow \pi^*$ transition with the experiment. The corresponding unshifted spectrum is enclosed as Fig. 1 of the supplementary material. Scalar-relativistic effects were taken into account using the spin-free exact two-component theory in its one-electron variant.

The first visible feature in the experimental spectrum, albeit a weak one, appears at round 287.02 eV and is assigned to the methyl carbon $1s \rightarrow \sigma^*$ transition [the corresponding DDNO is plotted in Fig. 4-(B)]. The computed transition energies for the $1s \rightarrow 3p$ transitions of the methyl carbon edge are very close to that of the carbonyl carbon $1s \rightarrow \pi^*$ transition [Fig. 4-(C,D)]. The $1s \rightarrow 4p$ transitions are around 1 eV higher and have much lower intensities. The transitions from 1s electron of the CH_2 carbon to a C-H σ^* orbital [Fig. 4-(E)], a C-O σ^* orbital [Fig. 4-(F)], and the 3p orbitals [Fig. 4-(G)] are located at 289.0 eV, 289.46 eV, and 289.73 eV, respectively. They also contribute to the intense feature centered around 288.3 eV. Interestingly, the most intense features from the CH_3 , CH_2 , and COO carbon edges all pile up in the pre-methyl carbon edge region ranging from 287 eV to 291 eV. As mentioned in the introduction, this renders the NEXAFS spectrum of the ESCA molecule substantially more complicated than the corresponding photoelectron spectrum.

The features in the experimental spectrum with transition energies higher than

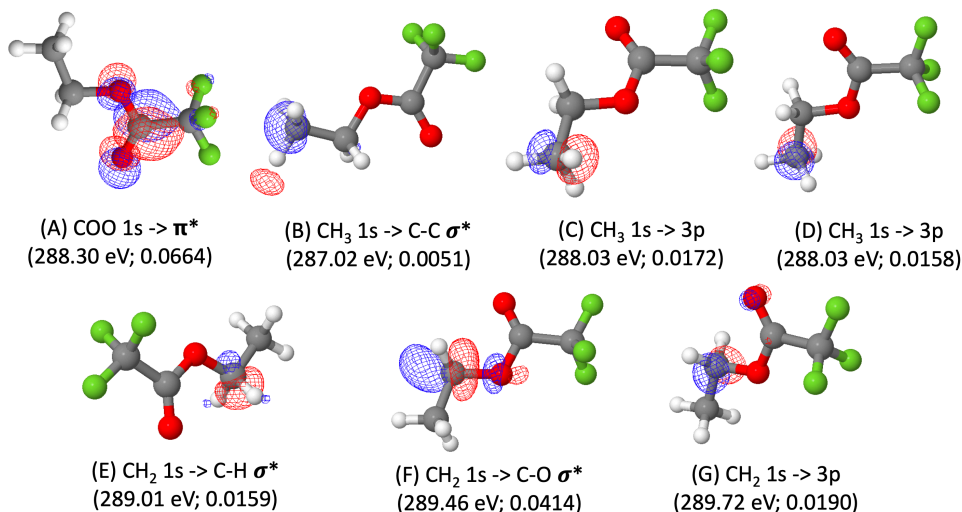


Figure 4: Density-difference natural virtual orbitals for core-excited states of COO, CH₃, and CH₂ C edges of the ESCA molecule that are the most intense ranging from 287 eV to 292 eV in the NEXAFS spectrum. Computed transition energies (with a 0.5 eV red shift) and oscillator strengths as used in Fig. 3 are enclosed in the parentheses.

293 eV are assigned to transitions of the CF₃ carbon edge. The transition around 293.5 eV is a valence excitation into a C-C π orbital, as shown in Fig. 5-(H). Interestingly, since the electron density between the two carbon atoms is heavily depleted due to the presence of three fluorine atoms with high electronegativity, this excitation that increases the electron density in an orbital of bonding character in this region produces an excited state with substantially lower energy than excitations to anti-bonding or Rydberg orbitals. The strong transition around 295.8 eV in the experimental spectrum is assigned to a valence excitation into the C-F σ^* orbital, which is calculated to be located at 296.14 eV [Fig. 5-(I)]. The broad feature centered around 297 eV consists of Rydberg transitions to 3p [Fig. 5-(J, K, L)] and 4p orbitals.

4.2. Remarks on the accuracy of the computed spectrum

Since computations play an important role in the assignment of the experimental spectrum, it is of interest to analyze the accuracy of the present computational results in more detail. First we focus on the description of core-excited Rydberg states. The use of the cc-pVTZ+4spd basis set for the target carbon can describe three sets of Rydberg states accurately, which is expected to cover important features in the experimental NEXAFS spectrum. As shown in Fig. 2-4 in the supplementary material, the cc-pVTZ+2spd basis provides an essentially converged description for valence excitations and the first sets of Rydberg excitations. However, inclusion of additional diffuse functions in the cc-pVTZ+3spd and cc-pVTZ+4spd sets provides richer structures and smaller intensities for higher Rydberg states, which clearly improves the quality of the

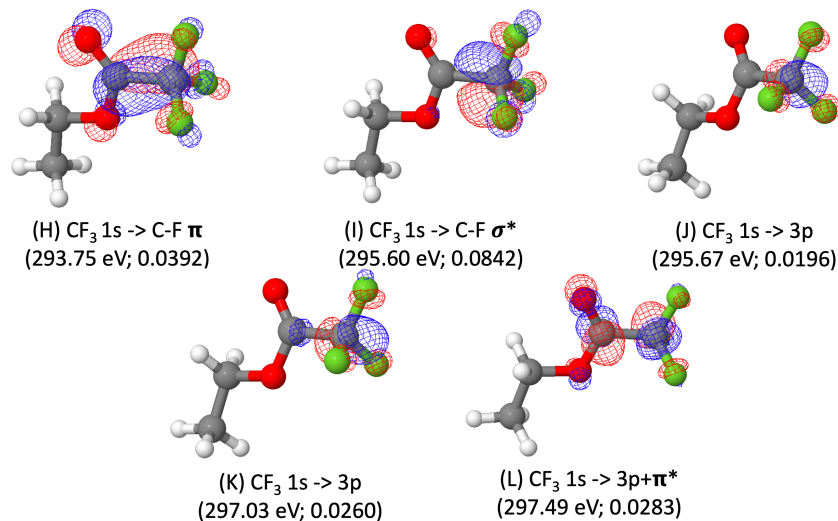


Figure 5: Density-difference natural virtual orbitals for intense core excited states of the CF_3 carbon edge of the ESCA molecule that are the most intense in the range of 292-299 eV in the NEXAFS spectrum. Computed transition energies (with a 0.5 eV red shift) and oscillator strengths as used in Fig. 3 are enclosed in the parentheses.

simulated spectrum. Inspection of the convergence of computational results shows that further addition of a fifth set of diffuse functions is unlikely to make a significant difference. It can thus be safely concluded that the treatment of the diffuse character of Rydberg states has been converged in the calculations using the cc-pVTZ+4spd basis set. Note that the augmentation of standard basis sets with sufficient diffuse functions is required to capture the diffuse nature of core-excited Rydberg states in general. In recent time-dependent density-functional theory (TDDFT) and restricted active space second order perturbation theory (RASPT2) calculations of the NEXAFS spectrum for the ESCA molecule [20], the use of standard basis sets also seem to lead to overestimation of intensities for Rydberg states. The present edge-specific approach is expected to be applicable for quantum-chemical methods other than coupled-cluster methods in a straightforward manner.

Accurate calculations of transition energies in the NEXAFS spectra has emerged as a major challenge. The experimental NEXAFS spectrum for the ESCA molecule presented here serves as a useful guide for benchmarking computational methods. Comparing the experimental and computational spectra, it appears that an accurate calculation for the separation between the $1s \rightarrow \pi^*$ transition of the carbonyl carbon and other transitions is a formidable task. Triples corrections clearly make important contributions here. Comparing the CCSD/cc-pVTZ+4spd spectrum (Fig. 4 of the supplementary material) and the CCSD(T)(a)^{*}/cc-pVTZ+4spd spectrum (Fig. 3), it can be seen that CCSD clearly overestimates the separation between the carbonyl carbon $1s \rightarrow \pi^*$ transition and the other transitions with higher energies. For example, the triples corrections red-shift the level positions of the transitions around 298 eV in

the spectrum by around 1 eV. A particularly difficult case is the separation between the carbonyl carbon $1s \rightarrow \pi^*$ transition and the CH_2 carbon $1s$ to C-O σ^* transition, which amounts to 1.41 eV at the CCSD level. The inclusion of triples corrections using CCSD(T)(a)* reduces this value to 1.16 eV. However, even CCSD(T)(a)* seems to overestimate this separation. In the experimental spectrum, only a small shoulder appears around 0.7 eV higher than the peak at 288.30 eV, which may indicate that the strong transition from the CH_2 carbon $1s$ orbital to the C-O σ^* orbital is located even closer to the carbonyl carbon $1s \rightarrow \pi^*$ transition, so that both of them contribute to the intense peak at 288.3 eV. This difficulty also persists for other computational methods. The TDDFT and RASPT2 calculations reported in Ref. [20] also successfully identified the carbonyl carbon $1s \rightarrow \pi^*$ transition as the first intense transition. However, the separation between the carbonyl carbon $1s \rightarrow \pi^*$ transition and the CH_2 carbon $1s$ to C-O σ^* transition has also been overestimated, with the TDDFT value being 1.5 eV and the RASSCF/RASPT2 value being as large as *ca.* 3 eV. It would be of interest to further investigate the calculations of these transition energies. In particular, delta-coupled-cluster methods, which directly calculate coupled-cluster wavefunctions for core excited states [74, 83, 49], may be able to provide more accurate energies for these states than the CVS-EOM-CC methods with approximate treatment of triple excitations. Further, vibronic coupling has been reported to make significant contributions to NEXAFS of CO_2 [82] and might be relevant to the ESCA molecule. This, however, is only speculation at the moment.

The sum of oscillator strengths for transitions from CF_3 , COO , CH_2 , and CH_3 edges are 0.277, 0.075, 0.108, and 0.066. The CF_3 edge has substantially higher intensity for core excitations than other edges, perhaps because the depletion of the electron density around this carbon and the sp^3 hybridization of this carbon atom offers good accommodation of various valence and Rydberg excitations. The present simulation has not considered contributions from core ionizations to the intensities of the absorption spectrum. Although the main peaks in the absorption spectrum seem to originate from core excitations and core ionizations likely contribute as smooth background beyond each edge, it would be of interest to investigate the contributions from core ionization using the Dyson orbital technique [84, 85, 86, 46] to get a complete description for the absorption spectrum. We should also mention that CVS-EOM-CC methods used here work with wavefunctions of bound states and are suitable for obtaining valence excitations and Rydberg excitations series that systematically approach the ionization threshold. They may not be suitable for describing resonance states beyond the ionization threshold, which in general requires the use of the so-called "non-Hermitian quantum mechanics" [87, 88]. Therefore, we only show results for core excited states with excitation energies below the ionization energy for every carbon edge in present discussions.

5. Summary and Outlook

We present the experimental photoabsorption spectrum of the ESCA molecule, ethyl trifluoroacetate, near the carbon K-edge and provide a theoretical interpretation of its spectral features. The theoretical methods advance the theoretical state-of-the-art by saturating basis-set effects to describe properly the diffuse Rydberg orbitals using a simple edge-specific approach for systematically expanding standard basis sets, and, by including triple excitations to capture the valence (in particular $1s \rightarrow \pi^*$)-to-Rydberg

energy intervals. These methods were benchmarked on other simple molecules (NH_3 , H_2CO , CO_2) and are thought to be generalizable. While most time-resolved NEXAFS spectra have been used to interpret dynamics due to optical/UV excitation of valence states [26, 16], NEXAFS can also provide fingerprints to follow inner-shell-initiated molecular dynamics - a discipline which is now feasible and a natural extension to x-ray pump/probe methods at XFELs [17, 89, 90, 91, 92]. Ethyl trifluoroacetate has already proven itself a model for ESCA (electron spectroscopy for chemical analysis), and, now extends its reach to advance theoretical methods to describe complex inner-shell photoabsorption spectra.

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References

- [1] Emma P, Akre R, Arthur J, Bionta R, Bostedt C, Bozek J, Brachmann A, Bucksbaum P, Coffee R, Decker F J *et al.* 2010 *Nature Photonics* **4** 641
- [2] Ishikawa T, Aoyagi H, Asaka T, Asano Y, Azumi N, Bizen T, Ego H, Fukami K, Fukui T, Furukawa Y, Goto S, Hanaki H, Hara T, Hasegawa T, Hatsui T, Higashiya A, Hirono T, Hosoda N, Ishii M, Inagaki T, Inubushi Y, Itoga T, Joti Y, Kago M, Kameshima T, Kimura H, Kirihara Y, Kiyomichi A, Kobayashi T, Kondo C, Kudo T, Maesaka H, Maréchal X M, Masuda T, Matsubara S, Matsumoto T, Matsushita T, Matsui S, Nagasono M, Nariyama N, Ohashi H, Ohata T, Ohshima T, Ono S, Otake Y, Saji C, Sakurai T, Sato T, Sawada K, Seike T, Shirasawa K, Sugimoto T, Suzuki S, Takahashi S, Takebe H, Takeshita K, Tamasaku K, Tanaka H, Tanaka R, Tanaka T, Togashi T, Togawa K, Tokuhisa A, Tomizawa H, Tono K, Wu S, Yabashi M, Yamaga M, Yamashita A, Yanagida K, Zhang C, Shintake T, Kitamura H and Kumagai N 2012 *Nature Photonics* **6** 540–544 URL <https://doi.org/10.1038/nphoton.2012.141>
- [3] Allaria E, Badano L, Bassanese S, Capotondi F, Castronovo D, Cinquegrana P, Danailov M B, D’Auria G, Demidovich A, De Monte R, De Ninno G, Di Mitri S, Diviacco B, Fawley W M, Ferianis M, Ferrari E, Gaio G, Gauthier D, Giannessi L, Iazzourene F, Kurdi G, Mahne N, Nikolov I, Parmigiani F, Penco G, Raimondi L, Rebernik P, Rossi F, Roussel E, Scafuri C, Serpico C, Sigalotti P, Spezzani C, Svandrlík M, Svetina C, Tróvó M, Veronese M, Zangrando D and Zangrando M 2015 *Journal of Synchrotron Radiation* **22** 485–491 URL <https://doi.org/10.1107/S1600577515005366>
- [4] Ko I, Kang H S, Heo H, Kim C, Kim G, Min C K, Yang H, Baek S, Choi H J, Mun G and et al 2017 *Applied Sciences* **7** 479 ISSN 2076-3417 URL <http://dx.doi.org/10.3390/app7050479>
- [5] Milne C, Schietinger T, Aiba M, Alarcon A, Alex J, Anghel A, Arsov V, Beard C, Beaud P, Bettoni S and et al 2017 *Applied Sciences* **7** 720 ISSN 2076-3417 URL <http://dx.doi.org/10.3390/app7070720>
- [6] Decking W, Abeghyan S, Abramian P, Abramsky A, Aguirre A, Albrecht C, Alou P, Altarelli M, Altmann P, Amyan K *et al.* 2020 *Nature Photonics* 1–7
- [7] Schoenlein R, Elsaesser T, Holldack K, Huang Z, Kapteyn H, Murnane M and Woerner M 2019 *Philosophical Transactions of the Royal Society A: Mathematical, Physical and Engineering Sciences* **377** 20180384 (Preprint <https://royalsocietypublishing.org/doi/pdf/10.1098/>

- rsta.2018.0384) URL <https://royalsocietypublishing.org/doi/abs/10.1098/rsta.2018.0384>
- [8] Popmintchev T, Chen M C, Popmintchev D, Arpin P, Brown S, Ališauskas S, Andriukaitis G, Balčiūnas T, Mücke O D, Pugzlys A, Baltuška A, Shim B, Schrauth S E, Gaeta A, Hernández-García C, Plaja L, Becker A, Jaron-Becker A, Murnane M M and Kapteyn H C 2012 *Science* **336** 1287–1291 ISSN 0036-8075 (Preprint <https://science.sciencemag.org/content/336/6086/1287.full.pdf>) URL <https://science.sciencemag.org/content/336/6086/1287>
- [9] Young L, Ueda K, Gühr M, Bucksbaum P H, Simon M, Mukamel S, Rohringer N, Prince K C, Masciovecchio C, Meyer M, Rudenko A, Rolles D, Bostedt C, Fuchs M, Reis D A, Santra R, Kapteyn H, Murnane M, Ibrahim H, Légaré F, Vrakking M, Isinger M, Kroon D, Gisselbrecht M, L’Huillier A, Wörner H J and Leone S R 2018 *Journal of Physics B: Atomic, Molecular and Optical Physics* **51** 032003 URL <https://doi.org/10.1088/2F1361-6455/2Faa9735>
- [10] Marinelli A, MacArthur J, Emma P, Guetg M, Field C, Kharakh D, Lutman A A, Ding Y and Huang Z 2017 *Applied Physics Letters* **111** 151101 (Preprint <https://doi.org/10.1063/1.4990716>) URL <https://doi.org/10.1063/1.4990716>
- [11] Huang S, Ding Y, Feng Y, Hemsing E, Huang Z, Krzywinski J, Lutman A A, Marinelli A, Maxwell T J and Zhu D 2017 *Phys. Rev. Lett.* **119**(15) 154801 URL <https://link.aps.org/doi/10.1103/PhysRevLett.119.154801>
- [12] Duris J, Li S, Driver T, Champenois E G, MacArthur J P, Lutman A A, Zhang Z, Rosenberger P, Aldrich J W, Coffee R *et al.* 2020 *Nature Photonics* **14** 30–36
- [13] Lutman A A, Coffee R, Ding Y, Huang Z, Krzywinski J, Maxwell T, Messerschmidt M and Nuhn H D 2013 *Phys. Rev. Lett.* **110**(13) 134801 URL <https://link.aps.org/doi/10.1103/PhysRevLett.110.134801>
- [14] Marinelli A, Ratner D, Lutman A A, Turner J, Welch J, Decker F J, Loos H, Behrens C, Gilevich S, Miahnahri A A, Vetter S, Maxwell T J, Ding Y, Coffee R, Wakatsuki S and Huang Z 2015 *Nature Communications* **6** 6369 URL <https://doi.org/10.1038/ncomms7369>
- [15] Lutman A A, Maxwell T J, MacArthur J P, Guetg M W, Berrah N, Coffee R N, Ding Y, Huang Z, Marinelli A, Moeller S *et al.* 2016 *Nature Photonics* **10** 745–750
- [16] Wolf T J A, Myhre R H, Cryan J P, Coriani S, Squibb R J, Battistoni A, Berrah N, Bostedt C, Bucksbaum P, Coslovich G, Feifel R, Gaffney K J, Grilj J, Martinez T J, Miyabe S, Moeller S P, Mücke M, Natan A, Obaid R, Osipov T, Plekan O, Wang S, Koch H and Gühr M 2017 *Nature Communications* **8** 29 URL <https://doi.org/10.1038/s41467-017-00069-7>
- [17] Picón A, Lehmann C S, Bostedt C, Rudenko A, Marinelli A, Osipov T, Rolles D, Berrah N, Bomme C, Bucher M, Doumy G, Erk B, Ferguson K R, Gorkhover T, Ho P J, Kanter E P, Krässig B, Krzywinski J, Lutman A A, March A M, Moonshiram D, Ray D, Young L, Pratt S T and Southworth S H 2016 *Nature Communications* **7** 11652 URL <https://doi.org/10.1038/ncomms11652>
- [18] Neville S P, Averbukh V, Patchkovskii S, Ruberti M, Yun R, Chergui M, Stolow A and Schuurman M S 2016 *Faraday Discuss.* **194**(0) 117–145 URL <http://dx.doi.org/10.1039/C6FD00117C>
- [19] Mukamel S, Healion D, Zhang Y and Biggs J D 2013 *Annual Review of Physical Chemistry* **64** 101–127 PMID: 23245522 (Preprint <https://doi.org/10.1146/annurev-physchem-040412-110021>) URL <https://doi.org/10.1146/annurev-physchem-040412-110021>
- [20] Nenov A, Segatta F, Bruner A, Mukamel S and Garavelli M 2019 *The Journal of Chemical Physics* **151** 114110 (Preprint <https://doi.org/10.1063/1.5116699>) URL <https://doi.org/10.1063/1.5116699>
- [21] Siegbahn K 1967 *Nova Acta Regiae Societatis Scientiarum Upsaliensis*
- [22] Siegbahn K 1970 *ESCA applied to free molecules* (North-Holland Pub. Co.)
- [23] Travnikova O, Børve K J, Patanen M, Söderström J, Miron C, Sæthre L J, Mårtensson N and Svensson S 2012 *Journal of Electron Spectroscopy and Related Phenomena* **185** 191 – 197 ISSN 0368-2048 special Issue in honor of Prof. T. Darrah Thomas: High-Resolution Spectroscopy of Isolated Species URL <http://www.sciencedirect.com/science/article/pii/S0368204812000552>
- [24] Inhester L, Oostenrijk B, Patanen M, Kokkonen E, Southworth S H, Bostedt C, Travnikova O, Marchenko T, Son S K, Santra R, Simon M, Young L and Sorensen S L 2018 *The Journal of Physical Chemistry Letters* **9** 1156–1163 PMID: 29444399 (Preprint <https://doi.org/10.1021/acs.jpcllett.7b03235>) URL <https://doi.org/10.1021/acs.jpcllett.7b03235>
- [25] Stöhr J 2003 *NEXAFS Spectroscopy* (Springer-Verlag Berlin Heidelberg New York)
- [26] Bhattacharjee A and Leone S R 2018 *Accounts of Chemical Research* **51** 3203–3211 (Preprint <https://doi.org/10.1021/acs.accounts.8b00462>) URL <https://doi.org/10.1021/acs.accounts.8b00462>

- 1021/acs.accounts.8b00462
- [27] Pertot Y, Schmidt C, Matthews M, Chauvet A, Huppert M, Svoboda V, von Conta A, Tehlar A, Baykusheva D, Wolf J P and Wörner H J 2017 *Science* ISSN 0036-8075 (Preprint <https://science.sciencemag.org/content/early/2017/01/04/science.aah6114.full.pdf>) URL <https://science.sciencemag.org/content/early/2017/01/04/science.aah6114>
- [28] Norman P and Dreuw A 2018 *Chemical Reviews* **118** 7208–7248 pMID: 29894157 (Preprint <https://doi.org/10.1021/acs.chemrev.8b00156>) URL <https://doi.org/10.1021/acs.chemrev.8b00156>
- [29] Besley N A 2020 *Accounts of Chemical Research* **53** 1306–1315 URL <https://doi.org/10.1021/acs.accounts.0c00171>
- [30] Stanton J F and Bartlett R J 1993 *J. Chem. Phys.* **98** 7029–7039
- [31] Krylov A I 2008 *Ann. Rev. Phys. Chem.* **59** 433–462
- [32] Besley N A 2012 *Chemical Physics Letters* **542** 42–46 ISSN 00092614 URL <http://dx.doi.org/10.1016/j.cplett.2012.05.059>
- [33] Coriani S, Christiansen O, Fransson T and Norman P 2012 *Phys. Rev. A* **85** 022507 ISSN 10502947
- [34] Southworth S H, Wehlitz R, Picón A, Lehmann C S, Cheng L and Stanton J F 2015 *J. Chem. Phys.* **142** ISSN 00219606
- [35] Peng B, Lestrangé P J, Goings J J, Caricato M and Li X 2015 *J. Chem. Theory Comput.* **11** 4146–4153 ISSN 15499626
- [36] Nascimento D R and Deprince A E 2017 *Journal of Physical Chemistry Letters* **8** 2951–2957 ISSN 19487185
- [37] Cederbaum L S, Domcke W and Schirmer J 1980 *Phys. Rev. A At., Mol., Opt. Phys.* **22** 206–222
- [38] Coriani S and Koch H 2015 *Journal of Chemical Physics* **143** ISSN 00219606 URL <http://dx.doi.org/10.1063/1.4935712>
- [39] Myhre R H, Coriani S and Koch H 2016 *Journal of Chemical Theory and Computation* **12** 2633–2643 ISSN 15499626
- [40] Yang Z, Schnorr K, Bhattacharjee A, Lefebvre P L, Epshtein M, Xue T, Stanton J F and Leone S R 2018 *J. Am. Chem. Soc.* **140** 13360–13366 URL <https://doi.org/10.1021/jacs.8b08303>
- [41] Vidal M L, Feng X, Epifanovsky E, Krylov A I and Coriani S 2019 *Journal of Chemical Theory and Computation* **15** 3117–3133 ISSN 15499626
- [42] Liu J, Matthews D, Coriani S and Cheng L 2019 *J. Chem. Theory Comput.* **15** 1642–1651
- [43] Frati F, De Groot F, Cerezo J, Santoro F, Cheng L, Faber R and Coriani S 2019 *Journal of Chemical Physics* **151** ISSN 00219606 URL <https://doi.org/10.1063/1.5097650>
- [44] Tenorio B N C, Moitra T, Nascimento M A C, Rocha A B and Coriani S 2019 *Journal of Chemical Physics* **150** ISSN 00219606 URL <http://dx.doi.org/10.1063/1.5096777>
- [45] Park Y C, Perera A and Bartlett R J 2019 *J. Chem. Phys.* **151** 164117 ISSN 0021-9606 URL <https://doi.org/10.1063/1.5117841>
- [46] Vidal M L, Krylov A I and Coriani S 2020 *Phys. Chem. Chem. Phys.* **22** 2693–2703 ISSN 1463-9076 URL <http://dx.doi.org/10.1039/C9CP03695D>
- [47] Faber R and Coriani S 2020 *Phys. Chem. Chem. Phys.* **22** 2642–2647 URL <http://dx.doi.org/10.1039/C9CP03696B>
- [48] Nanda K D, Vidal M L, Faber R, Coriani S and Krylov A I 2020 *Physical Chemistry Chemical Physics* **22** 2629–2641 ISSN 14639076
- [49] Matthews D A 2020 *Mol. Phys.* 1–8 ISSN 0026-8976 URL <https://doi.org/10.1080/00268976.2020.1771448>
- [50] Bazante A P, Perera A and Bartlett R J 2017 *Chem. Phys. Lett.* **683** 68–75 ISSN 0009-2614 URL <http://www.sciencedirect.com/science/article/pii/S0009261417304505>
- [51] Myhre R H, Coriani S and Koch H 2019 *Journal of Physical Chemistry A* **123** 9701–9711 ISSN 15205215
- [52] Tsuru S, Vidal M L, Pápai M, Krylov A I, Möller K B and Coriani S 2019 *Journal of Chemical Physics* **151** ISSN 00219606 URL <https://doi.org/10.1063/1.5115154>
- [53] Kjellsson L, Nanda K D, Rubensson J E, Doumy G, Southworth S H, Ho P J, March A M, Al Haddad A, Kumagai Y, Tu M F, Schaller R D, Debnath T, Bin Mohd Yusof M S, Arnold C, Schlotter W F, Moeller S, Coslovich G, Koralek J D, Minitti M P, Vidal M L, Simon M, Santra R, Loh Z H, Coriani S, Krylov A I and Young L 2020 *Physical Review Letters* **124** 236001 ISSN 0031-9007 URL <https://doi.org/10.1103/PhysRevLett.124.236001>
- [54] Myhre R H, Wolf T J, Cheng L, Nandi S, Coriani S, Gühr M and Koch H 2018 *Journal of Chemical Physics* **148** ISSN 00219606 URL <http://dx.doi.org/10.1063/1.5011148>
- [55] Southworth S H, Dunford R W, Ray D, Kanter E P, Doumy G, March A M, Ho P J, Krässig B, Gao Y, Lehmann C S, Picón A, Young L, Walko D A and Cheng L 2019 *Phys. Rev. A* **100**

- 22507 URL <https://link.aps.org/doi/10.1103/PhysRevA.100.022507>
- [56] Fano U and Cooper J W 1968 *Rev. Mod. Phys.* **40**(3) 441–507 URL <https://link.aps.org/doi/10.1103/RevModPhys.40.441>
- [57] Dehmer J L and Dill D 1975 *Phys. Rev. Lett.* **35**(4) 213–215 URL <https://link.aps.org/doi/10.1103/PhysRevLett.35.213>
- [58] Becker U and Shirley D A E 1996 *VUV and Soft X-ray Photoionization* (Plenum Press, New York and London)
- [59] Young L, Kanter E P, Krässig B, Li Y, March A M, Pratt S T, Santra R, Southworth S H, Rohringer N, DiMauro L F, Doumy G, Roedig C A, Berrah N, Fang L, Hoener M, Bucksbaum P H, Cryan J P, Ghimire S, Glowia J M, Reis D A, Bozek J D, Bostedt C and Messerschmidt M 2010 *Nature* **466** 56–61 URL <https://doi.org/10.1038/nature09177>
- [60] Santra R and Young L 2015 Interaction of intense x-ray beams with atoms *Synchrotron Light Sources and Free-Electron Lasers* ed E Jaeschke S Khan J R S J B H (Springer International Publishing) pp 1–24
- [61] Ueda K 2003 *Journal of Physics B: Atomic, Molecular and Optical Physics* **36** R1–R47 URL <https://doi.org/10.1088/0953-4075/36/2F36/2F4%2F201>
- [62] Miron C, Simon M, Leclercq N and Morin P 1997 *Review of Scientific Instruments* **68** 3728–3737 (Preprint <https://doi.org/10.1063/1.1148017>) URL <https://doi.org/10.1063/1.1148017>
- [63] Liu X J, Nicolas C, Robert E and Miron C 2014 *Journal of Physics: Conference Series* **488** 142005 URL <https://doi.org/10.1088/1742-6596/488/2F14/2F142005>
- [64] Nicolas C and Miron C 2012 *Journal of Electron Spectroscopy and Related Phenomena* **185** 267–272
- [65] Adachi J i, Kosugi N, Shigemasa E and Yagishita A 1996 *The Journal of Physical Chemistry* **100** 19783–19788 (Preprint <https://doi.org/10.1021/jp962025j>) URL <https://doi.org/10.1021/jp962025j>
- [66] Dylla K G 2001 *J. Chem. Phys.* **115** 9136–9143
- [67] Liu W and Peng D 2009 *J. Chem. Phys.* **131** 1–5
- [68] Woon D E and Dunning, Jr T H 1995 *J. Chem. Phys.* **103** 4572–4585
- [69] Matthews D A, Cheng L, Harding M E, Lipparini F, Stopkiewicz S, Jagau T C, Szalay P G, Gauss J and Stanton J F 2020 *J. Chem. Phys.* **152** 214108 ISSN 0021-9606 URL <https://doi.org/10.1063/5.0004837>
- [70] Stanton J F, Gauss J, Cheng L, Harding M E, Matthews D A and Szalay P G CFour, Coupled-Cluster techniques for Computational Chemistry, a quantum-chemical program package with contributions from A.A. Auer, R.J. Bartlett, U. Benedikt, C. Berger, D.E. Bernholdt, Y.J. Bomble, O. Christiansen, F. Engel, R. Faber, M. Heckert, O. Heun, M. Hilgenberg, C. Huber, T.-C. Jagau, D. Jonsson, J. Jusélius, T. Kirsch, K. Klein, W.J. Lauderdale, F. Lipparini, T. Metzroth, L.A. Mück, D.P. O’Neill, D.R. Price, E. Prochnow, C. Puzzarini, K. Ruud, F. Schiffmann, W. Schwalbach, C. Simmons, S. Stopkiewicz, A. Tajti, J. Vázquez, F. Wang, J.D. Watts and the integral packages MOLECULE (J. Almlöf and P.R. Taylor), PROPS (P.R. Taylor), ABACUS (T. Helgaker, H.J. Aa. Jensen, P. Jørgensen, and J. Olsen), and ECP routines by A. V. Mitin and C. van Wüllen. For the current version, see <http://www.cfour.de>.
- [71] Cheng L and Gauss J 2011 *J. Chem. Phys.* **135** 084114
- [72] Matthews D A, Gauss J and Stanton J F 2013 *J. Chem. Theory Comput.* **9** 2567–2572 ISSN 1549-9618
- [73] Matthews D A and Stanton J F 2015 *J. Chem. Phys.* **142** 064108
- [74] Zheng X and Cheng L 2019 *J. Chem. Theory Comput.* **15** 4945–4955
- [75] Krylov A I 2020 *J. Chem. Phys.* **submitted**
- [76] Head-Gordon M, Grana A M, Maurice D and White C A 1995 *J. Phys. Chem.* **99** 14261–14270 ISSN 00223654
- [77] Schirmer J, Trofimov A B, Randall K J, Feldhaus J, Bradshaw A M, Ma Y, Chen C T and Sette F 1993 *Phys. Rev. A* **47**(2) 1136–1147 URL <https://link.aps.org/doi/10.1103/PhysRevA.47.1136>
- [78] Kowalski K and Piecuch P 2001 *J. Chem. Phys.* **115** 643–651
- [79] Kucharski S A, Włoch M, Musiał M and Rodney J Bartlett R 2001 *J. Chem. Phys.* **115** 8263–8266
- [80] Matthews D A and Stanton J F 2016 *J. Chem. Phys.* **145** 124102
- [81] Remmers G, Domke M, Puschmann A, Mandel T, Xue C, Kaindl G, Hudson E and Shirley D A 1992 *Phys. Rev. A* **46**(7) 3935–3944 URL <https://link.aps.org/doi/10.1103/PhysRevA.46.3935>
- [82] Prince K C, Avaldi L, Coreno M, Camilloni R and de Simone M 1999 *Journal of Physics B: Atomic, Molecular and Optical Physics* **32** 2551–2567 URL <https://doi.org/10.1088>

- 2F0953-4075%2F32%2F11%2F307
- [83] Lee J, Small D W and Head-Gordon M 2019 *J. Chem. Phys.* **151** 214103 ISSN 0021-9606 URL <https://doi.org/10.1063/1.5128795>
 - [84] Fujikawa T 1982 *J. Phys. Soc. Japan* **51** 2619–2627 ISSN 0031-9015 URL <https://doi.org/10.1143/JPSJ.51.2619>
 - [85] Lindgren I 2004 *J. Electron Spectros. Relat. Phenomena* **137-140** 59–71 ISSN 0368-2048 URL <http://www.sciencedirect.com/science/article/pii/S0368204804000374>
 - [86] Neville S P, Averbukh V, Ruberti M, Yun R, Patchkovskii S, Chergui M, Stolow A and Schuurman M S 2016 *J. Chem. Phys.* **145** 144307 ISSN 0021-9606 URL <https://doi.org/10.1063/1.4964369>
 - [87] Moiseyev N 2011 *Non-Hermitian Quantum Mechanics* (Cambridge: Cambridge Univ. Press)
 - [88] Jagau T C, Bravaya K B and Krylov A I 2017 *Annu. Rev. Phys. Chem.* **68** 525–553 URL <https://doi.org/10.1146/annurev-physchem-052516-050622>
 - [89] Zheng X, Liu J, Doumy G, Young L and Cheng L 2020 *The Journal of Physical Chemistry A* **124** 4413–4426 pMID: 32255349 (*Preprint* <https://doi.org/10.1021/acs.jpca.0c00901>) URL <https://doi.org/10.1021/acs.jpca.0c00901>
 - [90] Liekhus-Schmaltz C E, Tenney I, Osipov T, Sanchez-Gonzalez A, Berrah N, Boll R, Bomme C, Bostedt C, Bozek J D, Carron S *et al.* 2015 *Nature communications* **6** 1–7
 - [91] Oberli S, González-Vázquez J, Rodríguez-Perelló E, Sodupe M, Martín F and Picón A 2019 *Physical Chemistry Chemical Physics* **21** 25626–25634
 - [92] Cooper B, Kolorenč P, Frasinski L J, Averbukh V and Marangos J P 2014 *Faraday discussions* **171** 93–111